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no: 66. 1. What colour do the following indicators turn when added to a base or alkali (such as sodium hydroxide)? (a) Methyl orange (b) litmus (c) red cabbage extract.  
Solutions: (a) When methyl orange is added to a base it turns colour into yellow (b) Litmus paper changes into blue when ...

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(glacial) Acetic Acid (25  
ML), And A Teflon  
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$\text{KMnO}_4$  convert alkenes into diols (glycols). This behaviour is also used as a qualitative test for the presence of double or triple bonds in a molecule, since the reaction decolorizes the initially purple permanganate solution and generates a brown precipitate ( $\text{MnO}_2$ ).

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No:66. Question 1:  
What colour do the  
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turn when added to a  
base or alkali (such as  
sodium hydroxide) ?  
(a) methyl orange (b)  
litmus (c) red cabbage  
extract Solution : (a)  
Yellow. (b) Blue. (c)  
Green. Question 2:

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