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In the Organic Chemistry Teaching labs at CU, two methods of drying solutions are commonly used: saturated aqueous sodium chloride and solid drying agents. These two methods are described below. Saturated Aqueous Sodium Chloride . The bulk of the water can often be removed by shaking or "washing" the organic layer with saturated aqueous sodium chloride (otherwise known as brine). The salt ...

Drying Organic Solutions - Organic Chemistry

In chemistry, a solution is a special type of homogeneous mixture composed of two or more substances. In such a mixture, a solute is a substance dissolved in another substance, known as a solvent. The mixing process of a solution happens at a scale where the effects of chemical polarity are involved, resulting in interactions that are specific to solvation.

Solution - Wikipedia

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Fischer Projections - Chemistry LibreTexts

With some solutions (e.g. dichloromethane), care should be taken to not shake too vigorously, as these solutions often form emulsions (where the interface between the solutions doesn't clarify). With solutions prone to emulsions, a funnel should be gently rocked for one minute. Place the separatory funnel upright in the ring clamp to allow the layers to fully separate. The interface between ...

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